

Half-term Five April – May

Year 10 Chemistry – Electrolysis and Metal Extraction

Lesson	Instructions	Resources	Curriculum
1	Electrolysis Work through the tasks from the pages in your booklet	Oak Academy Lesson Electrolysis of Molten Compounds PowerPoint Oak Academy Lesson Electrolysis Half Equations	Electrolysis
2	Electrolysis Core Practical I Work through the tasks from the pages in your booklet	Oak Academy Lesson Developing an Electrolysis Hypothesis PowerPoint	Electrolysis
3	Electrolysis Core Practical II Work through the tasks from the pages in your booklet	<u>PowerPoint</u>	Electrolysis
4	Products From Electrolysis Work through the tasks from the pages in your booklet	Oak Academy Lesson Electrolysis of Solutions PowerPoint	Electrolysis
5	Electrolysis Review Work through the tasks from the pages in your booklet	Oak Academy Lesson Electrolysis Review PowerPoint	Electrolysis
6	Reactivity Series Work through the tasks from the pages in your booklet	Oak Academy Lesson Investigating the Reactivity of Metals PowerPoint	Obtaining and Using Metals
7	Displacement Reactions Work through the tasks from the pages in your booklet	Oak Academy Lesson Displacement Reactions of Metals PowerPoint	Obtaining and Using Metals

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8	Extracting Metals Work through the tasks from the pages in your booklet	Oak Academy Lesson Extraction of Aluminium PowerPoint	MONKSEAT Obtaining and Using Metals
9	Biological Methods of Extraction – Higher Tier Only Work through the tasks from the pages in your booklet	<u>PowerPoint</u>	Higher Obtaining and Using Metals
10	Life Cycle Assessment and Recycling Work through the tasks from the pages in your booklet	Oak Academy Lesson Electrolysis Review PowerPoint	Obtaining and Using Metals
11	Dynamic Equilibrium Work through the tasks from the pages in your booklet	Oak Academy Lesson Reversible Reactions PowerPoint	Reversible Reactions and Equilibria
12	Changing Equilibrium – Higher Tier Only Work through the tasks from the pages in your booklet	Oak Academy Lesson Le Chatelier's Principle: Effect of Changing Concentration and Temperature PowerPoint Oak Academy Lesson Chatelier's Principle: Effect of Changing Pressure	Higher Reversible Reactions and Equilibria

Please email any queries and completed work to your Chemistry teacher.

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